

靜宜大學 104 學年度碩士班招生考試試題

學系：化粧品科學系

科目：有機化學

Part I (50%)

一、單選題 (30 pts, 3 pts/ each)

- Which atom is described by the electron configuration $1s^2 2s^2 2p^6 3s^2 3p^3$?
(A) phosphorus (B) oxygen (C) calcium (D) aluminum
- Which molecules are polar?
 CO_2 (I), CH_3Cl (II), CCl_4 (III), NH_3 (IV)
(A) I, III (B) II, III (C) II, IV (D) II, III, IV
- Which compound has ionic and covalent bonds?
(A) CH_3OH (B) KBr (C) HCl (D) NaHSO_4
- Which chemical bond is the most polar?
(A) O-H (B) C-H (C) F-F (D) N-H
- What is the relationship between these two structures?
$$\begin{array}{c} \text{NH}_2 \\ | \\ \text{CH}_3\text{CHCH}_3 \end{array} \quad \text{and} \quad \text{CH}_3\text{CH}_2\text{CH}_2\text{NH}_2$$

(A) diastereomers (B) enantiomers (C) conformational isomers (D) constitutional isomers
- Rank the following groups in order of increasing priority (least first)?
 $-\text{CH}(\text{CH}_3)_2$ (I), $-\text{OCH}_3$ (II), $-\text{CO}_2\text{H}$ (III), $-\text{CH}=\text{CH}_2$ (IV)
(A) I, II, III, IV (B) II, I, III, IV (C) I, IV, III, II (D) IV, I, III, II
- Which compound reacts most readily in the nitration reaction?
(A) benzene (B) phenol (C) nitrobenzene (D) benzoic acid
- Which compound has the strongest acidity?
(A) ethanol (B) phenol (C) *p*-nitrophenol (D) *p*-methylphenol
- Which compound is secondary alkyl halide?
(A) 2-bromo-2-methylpropane (B) 1-bromo-2-methylpropane
(C) 1-bromobutane (D) 2-bromobutane
- Which is the structural formula for heptanoic acid?
(A) $\text{CH}_3(\text{CH}_2)_4\text{CH}_2\text{OH}$ (B) $\text{CH}_3(\text{CH}_2)_5\text{CHO}$ (C) $\text{CH}_3(\text{CH}_2)_4\text{CO}_2\text{H}$ (D) $\text{CH}_3(\text{CH}_2)_5\text{CO}_2\text{H}$

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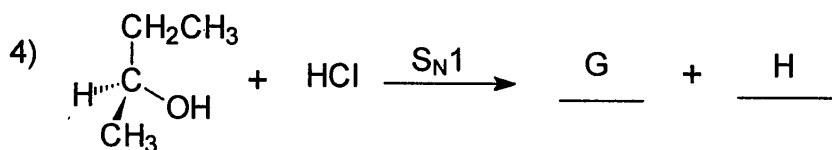
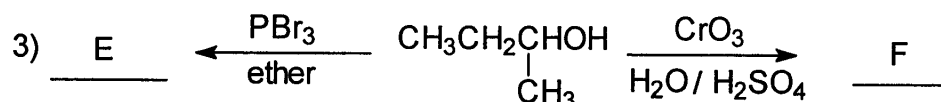
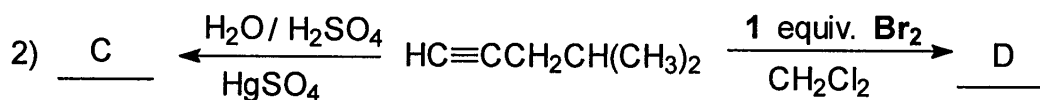
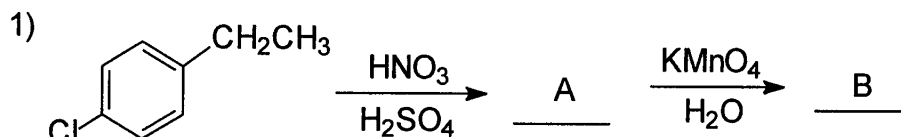
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科目：有機化學

Part I

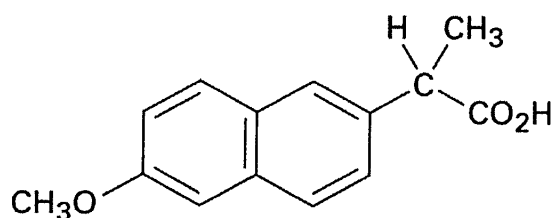
二、非選擇題

1. Please write the structural formula for the **major product** in each of the following reactions. (Note: Please give the correct stereochemistry, if necessary.) (20 pts, 5pts/each)

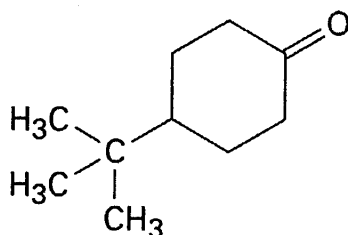


Part II (50%)

1. Calculate the number of sigma (σ) and pi (π) of the following compounds? (8%)



2. Calculate the number of primary carbon, secondary carbon, tertiary carbon and quaternary carbon of the following compound? (12%)



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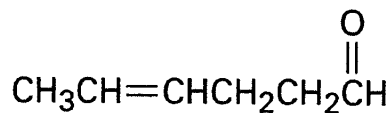
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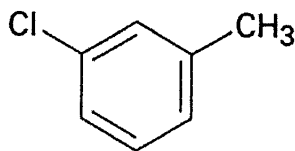
3. Give the IUPAC name of the following compound?

(16%)

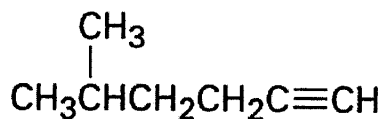
(i)



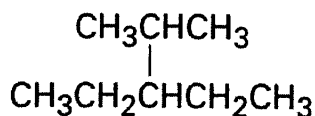
(ii) hint: Cl (chloro)



(iii)



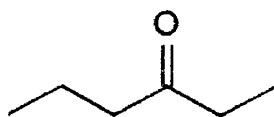
(iv)



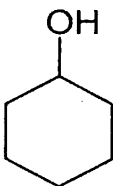
4. Which one of the following compounds is most consistent with the IR spectrum given in Figure-1? Explain your reasoning.

(6%, 8%)

(i)



(ii)



(iii)

